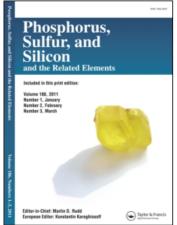
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## Microstructual Changes of Stainless Steel Surface by Hydroxyapatite and Metal Alternate Electrochemical Plating

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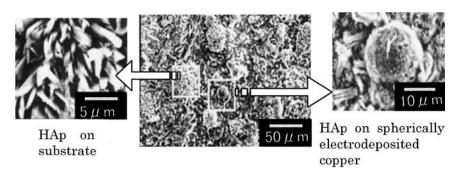
## MICROSTRUCTUAL CHANGES OF STAINLESS STEEL SURFACE BY HYDROXYAPATITE AND METAL ALTERNATE ELECTROCHEMICAL PLATING

Mikako Nakashima, Hiroko Shimizu, Hideki Monma, Satoshi Takahashi, Toshinori Okura, and Masao Kosaka Kogakuin University, Japan

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Calcium hydroxyapatite  $[Ca_{10}(PO_4)_6(OH)_2;HAp]$  coatings have been studied for a bioactive modification of stainless steel surface. HAp coatings are easily made on metallic substrates by electrolysis, however the coating layers become inevitably porous due to the generation of  $H_2$  gas. In this study, we report that the pores were filled with metals by the additional electroplating of the metals.

An electrolyte named  $E_A$  for HAp coating was prepared by dissolving  $Ca(H_2PO_4)_2 \cdot H_2O$ ,  $NaNO_3$ , NaF, and nitric acid, and maintained at  $80^{\circ}C$ . The working electrodes were two platinum sheets, and plating electrode was stainless steel. First, a constant direct current at  $10mA/cm^2$  was passed for 5 min through the  $E_A$  for the deposition of



**FIGURE 1** SEM micrographs of electrodeposited HAp-copper-HAp multilayered coating. (pH of  $E_{\rm A}=3.0$ ).

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HAp on the cathode substrate. Thus obtained HAp-coated substrates were treated as the cathode by electrolyzing solutions containing metal (M = Zn, Cu, or Fe) ions at  $25^{\circ}\text{C}$  under similar electrolysis conditions. Next, thus obtained cathode substrates were treated as the cathode by electrolyzing  $E_A$  again under the same conditions (Figure 1). The resulting HAp-metal-HAp multilayered coatings were characterized by SEM, XRD peak-separation analysis, and TF-XRD.